



Q-Chem Program: Best Available Fundamental Methods

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Q-Chem Inc.

Pittsburgh, PA

- All types of functionals – B3LYP, BMK, M06
- Linear scaling
- Dual basis
- Special methods: Constrained-DFT, spin-flip DFT

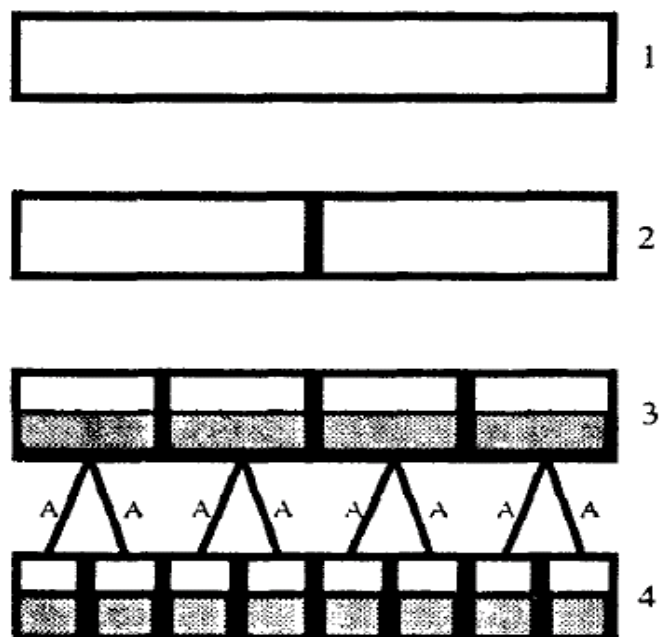
HF and DFT have three components:

- Coulomb $O(N^2)$:
$$F_{\mu\nu} \leftarrow \sum_{\lambda\sigma} (\mu\nu | r_{12}^{-1} | \lambda\sigma) P_{\lambda\sigma}$$
- Exchange $O(N^2)$:
$$F_{\mu\nu} \leftarrow \sum_{\lambda\sigma} (\mu\lambda | r_{12}^{-1} | \nu\sigma) P_{\lambda\sigma}$$
- DFT Exchange&Correlation $O(N)$:

$$F_{\mu\nu} \leftarrow \sum_{Atom} \sum_i f'_{\rho,i} \phi_{\mu,i} \phi_{\nu,i}$$

Developers: White, Head-Gordon @ Berkeley

- Invented by Greenard at MIT for point charges in 1987
- Q-Chem the first in the implementation to quantum chemistry for continuous charge in 1996
- First realization of linear-scaling DFT molecules



Developers: *Shao, Head-Gordon*

- Reduce the number of intermediates
- Faster coulomb energy and force

	<i>J-engine</i>	<i>integrals</i>	<i>speed-up</i>
(pp pp)	28.1	160.2	5.7
(dd dd)	69.3	792.3	11.4
(ff ff)	141.4	2630.3	18.6

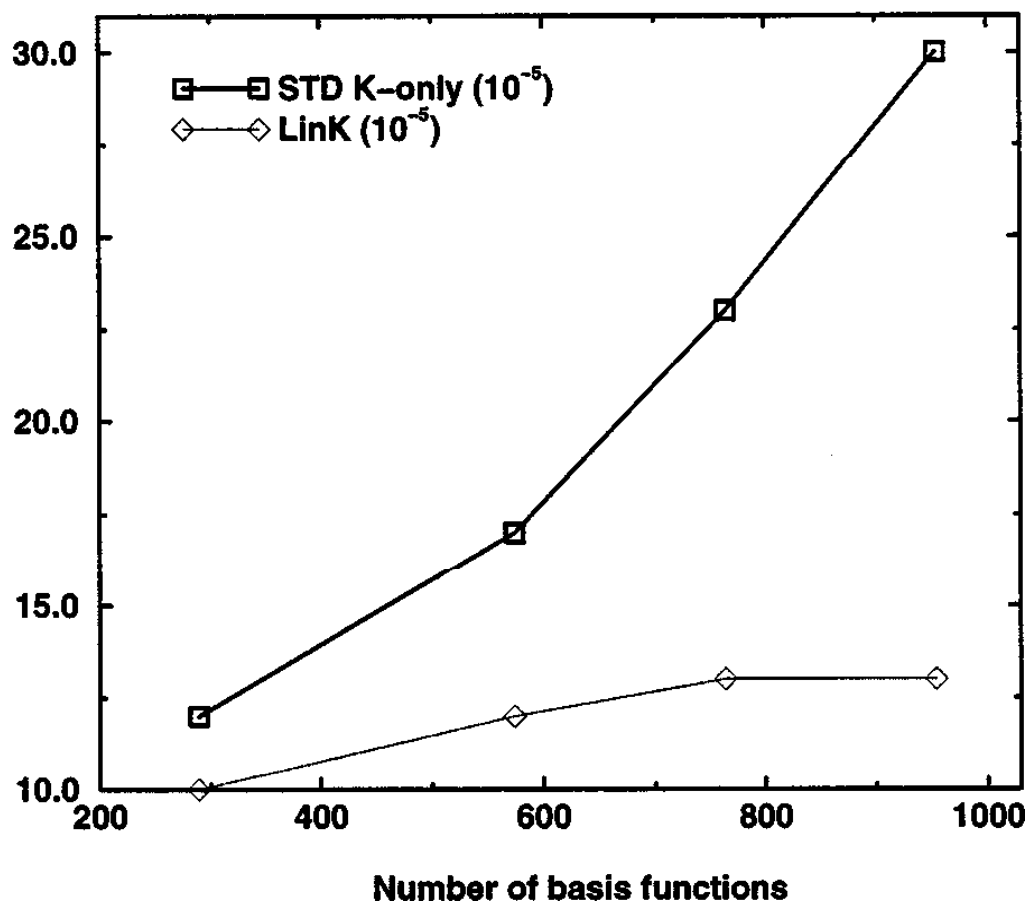
Ochsenfeld, Head-Gordon @berkeley

Using the sparsity
of density matrix:

$$\sum_{\lambda\sigma} (\mu\lambda | r_{12}^{-1} | \nu\sigma) P_{\lambda\sigma}$$

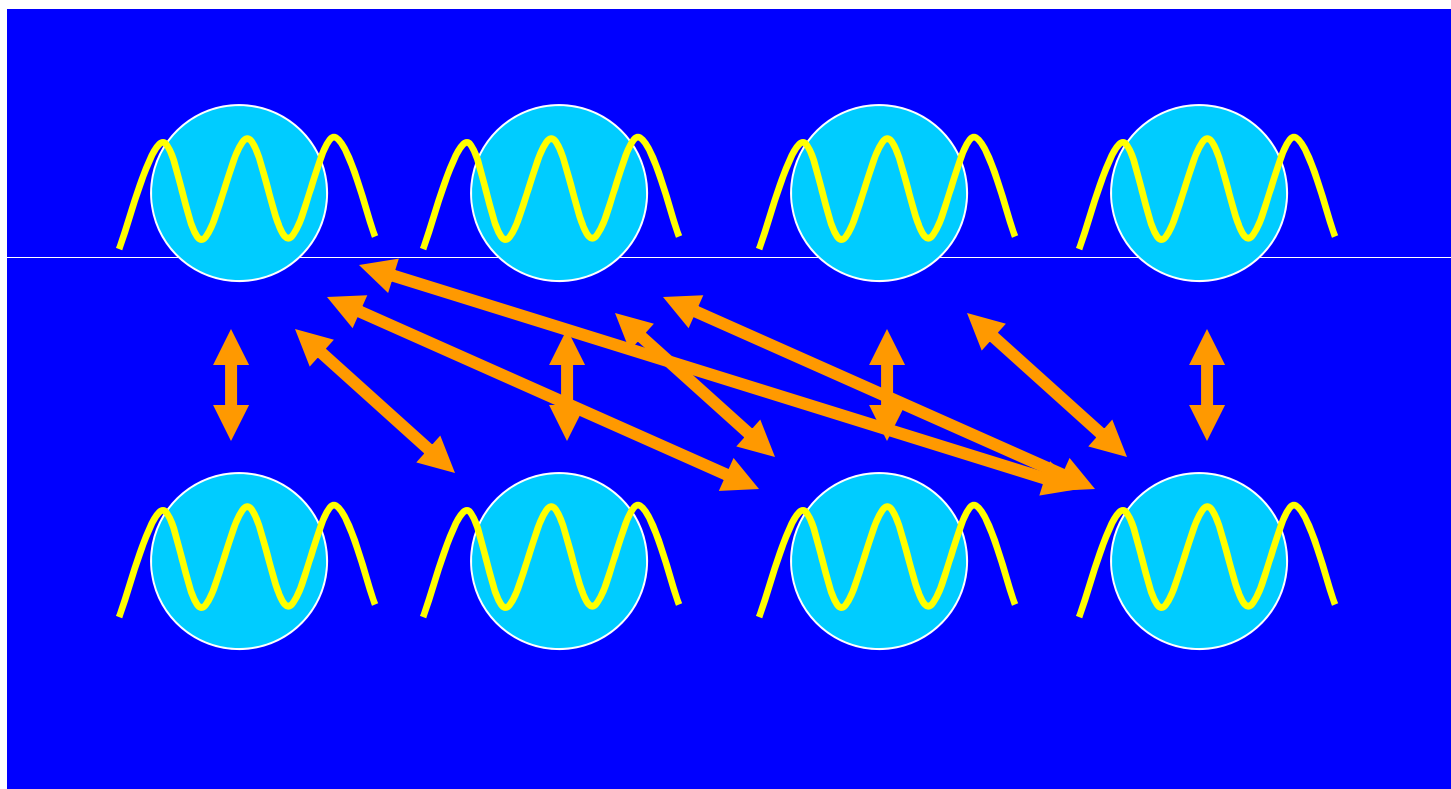
CPU [sec]

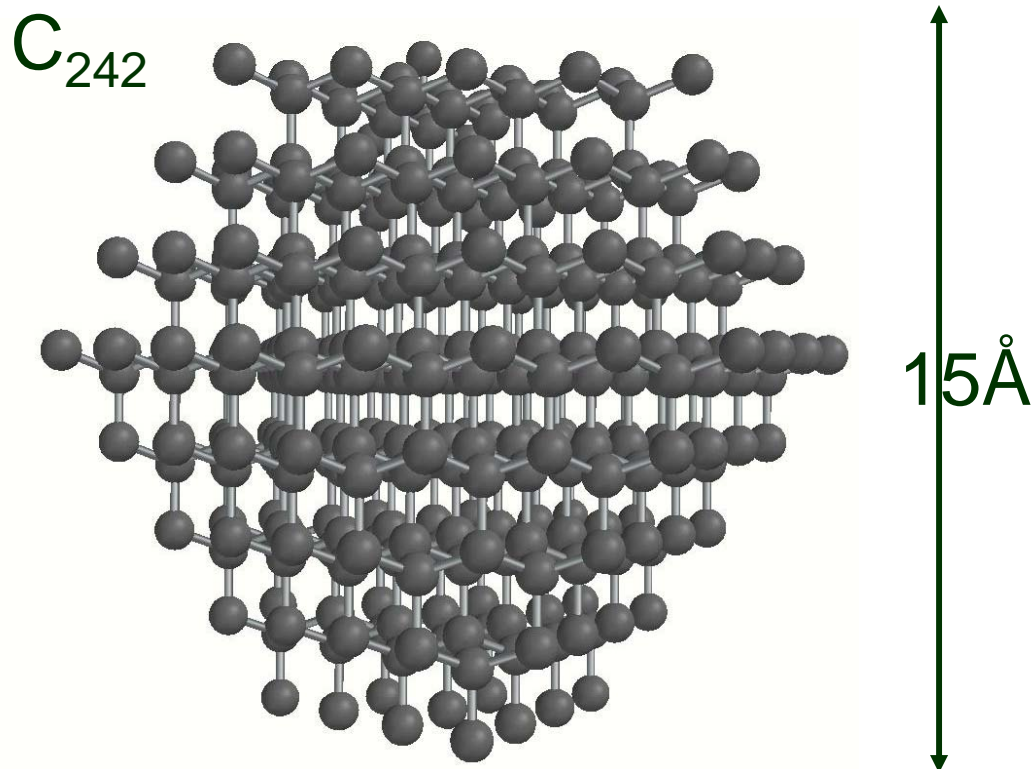
C_nH_{2n+2} (6-31G*)



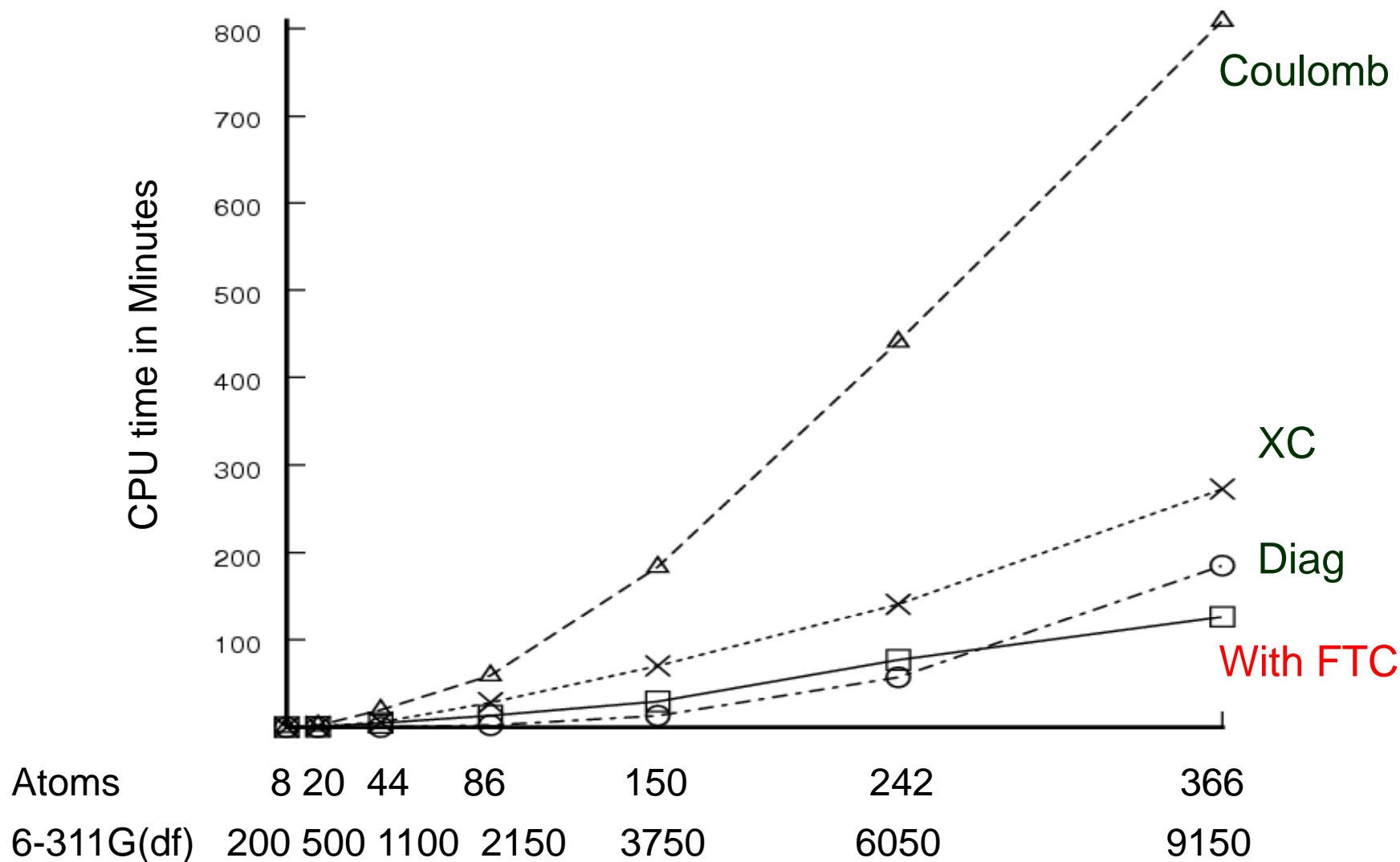
Fusti-Molnar, Kong

- Uses planewaves as auxiliary basis
- Fusti-Molnar and Peter Pulay's operator (JCP, Vol 116, 2002)
- Calculates the integrals with diffuse functions
- $O(N^2)$ in basis set size versus $O(N^4)$
- $O(N)$ in system size with a small crossover
- No loss of accuracy





6-311G(d), 6-311G(df), IBM Pwr4, 1 CPU



Liang, Steele, Head-Gordon @berkeley.edu

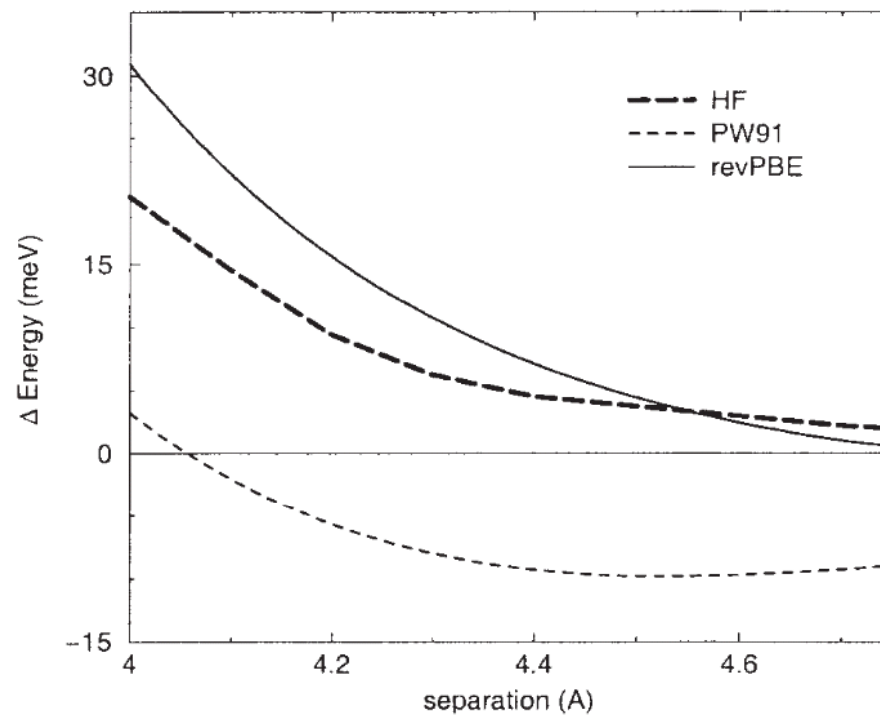
Large Basis Calculation at the Cost of Small Basis

- Basis extrapolation
- 10 times faster for SCF

G2 database Calculations with B3LYP in kcal/mol.
Target basis is always 6-311++G(3df,3pd).

Small basis	single basis	dual basis
6-311G	MAD = 24.3	MAD = 4.0
6-311G*	MAD = 7.0	MAD = 2.2
6-311G**	MAD = 4.9	MAD = 2.2
6-311++G(3df,3pd)	MAD = 2.2	MAD = 2.2

DFT Long Range Correction



Chai, Head-Gordon @berkeley.edu

$$\frac{1}{r_{12}} = \frac{\text{erf}(\omega r_{12})}{r_{12}} + \frac{\text{erfc}(\omega r_{12})}{r_{12}}$$

- Several versions
- ω B97X-D
- Much better than B3LYP-D:
Mean Absolute Error: 1.93 vs 2.77

Wu, van Hoorhis @mit.edu

Constrained DFT

- DFT calculation with predefined charge separation
- Used for Marcus Theory for CT reactions

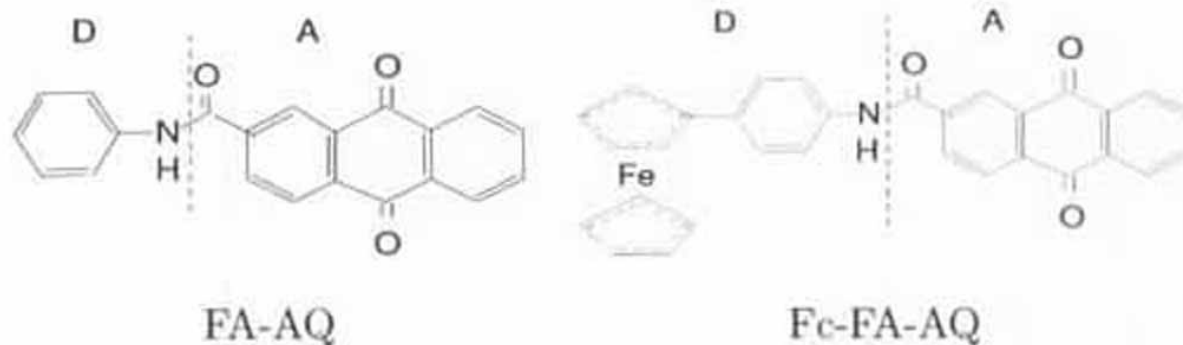
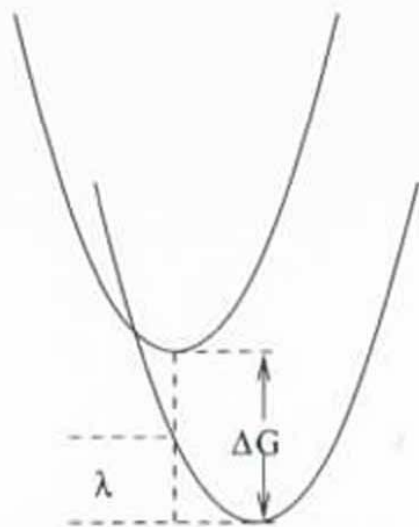


TABLE 2: Calculated Driving Force and Inner-Sphere Reorganization Energy for the Charge Recombination Reaction of FA-AQ and Fc-FA-AQ^a

	τ_{CR}	$-\Delta G_{CR}$ (exptl)	$-\Delta G_{CR}$ (calcd)		λ_i
			gas phase	in DMSO	
FA-AQ	>900 μ s	2.24	3.25	2.31	0.26
Fc-FA-AQ	20 ps	1.16	2.29	1.02	0.21

Dachsel, Korambath, Brown, Fusti-Molnar, Kong

- Up to 2nd derivatives for HF and DFT
- Including linear scaling J-engine, CFMM and FTC
- Distributed memory for frequency calc
- Available on all the platforms

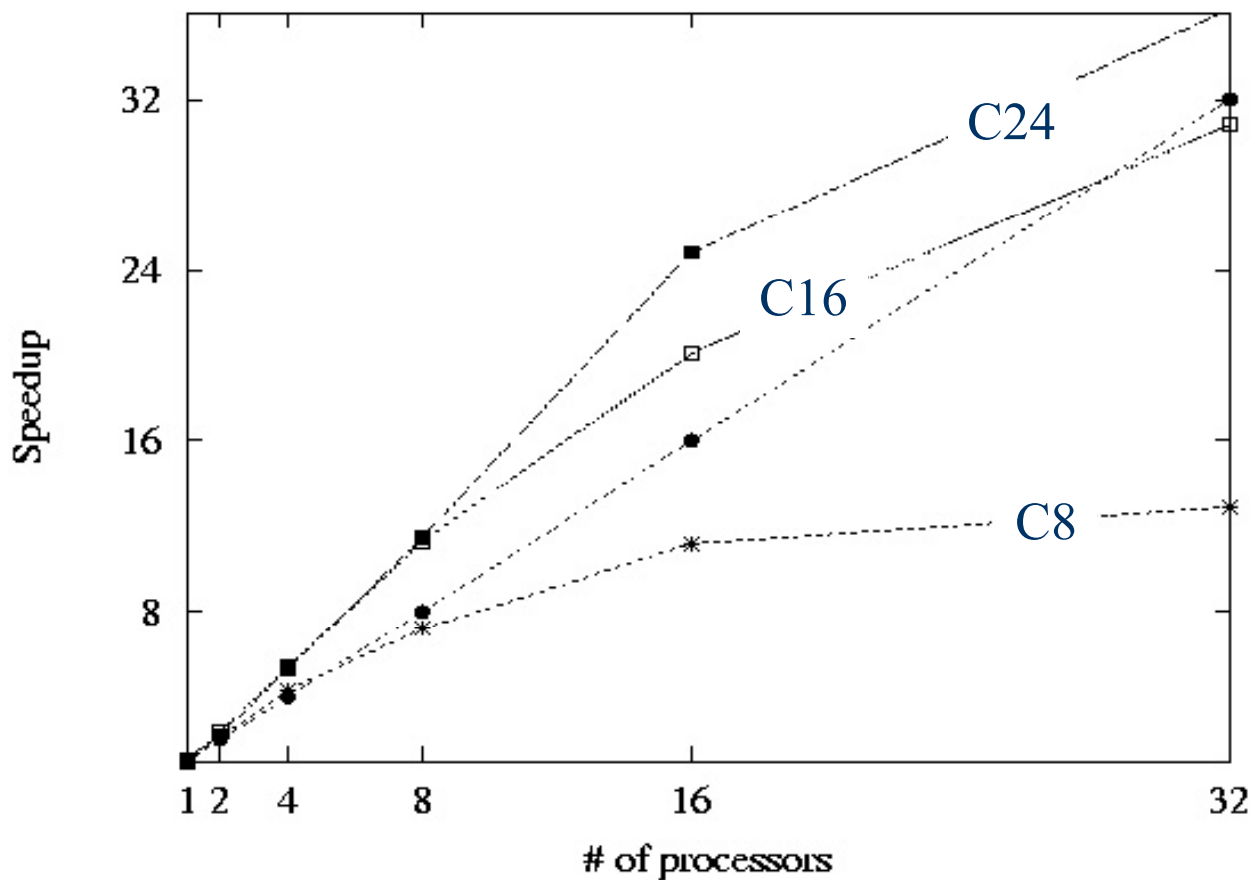
CPSCF: $(1 - A)\mathbf{B}^x = \mathbf{B}_0^x$

Parallelization Strategy:

Distribute the Nuclear Perturbations x

- All steps in parallel except (2)
- Scalable
 - the larger the system, the higher the speed-up
- Virtually no communication
- Memory usage distributed

CPHF speedup for alkane series in Q-CHEM on PC Linux cluster



- Very efficient semidirect MP2
- Local MP2
- RI-MP2
- Scaled modified-spin models
- Dual basis

$$\langle \mu\nu | \lambda\sigma \rangle \rightarrow \langle ij | kl \rangle$$



Faster DFT, MP2

Jung, Lochan, Dutoi, Head-Gordon @berkeley.edu

MP2 at HF Speed

RI approximation

$$\langle \mu\nu | \lambda\sigma \rangle \rightarrow \langle ij | kl \rangle$$

$$|\mu\nu\rangle \approx \sum_{K,L} |L\rangle V_{KL}^{-1} \langle K | \mu\nu \rangle$$

$$\langle \mu\nu | \lambda\sigma \rangle \rightarrow \sum_{K,L} \langle \mu\nu | K \rangle V_{KL}^{-1} \langle L | \lambda\sigma \rangle$$

MP2 at HF Speed

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- Speed-up: **7X – 10X**
- Errors are chemically insignificant:

Atomization energies of
148 neutral molecules (G2 set)

RMS = **0.10 kcal/mol**

Max = **0.27 kcal/mol**

Jung, Lochan, Dutoi, Head-Gordon @berkeley.edu

Empirical MP2 Models

How MP2 can be faster and more accurate?

$$E_{\text{SCS-MP2}} = c_{\text{OS}} E_{\text{MP2}}^{\text{OS}} + c_{\text{SS}} E_{\text{MP2}}^{\text{SS}}$$

- SOS (scaled opposite-spin)
- Faster: $O(N^4)$ vs $O(N^5)$
- Can be more accurate
- Wrong asymptotic behavior!

MOS – Modified Opposite Spin

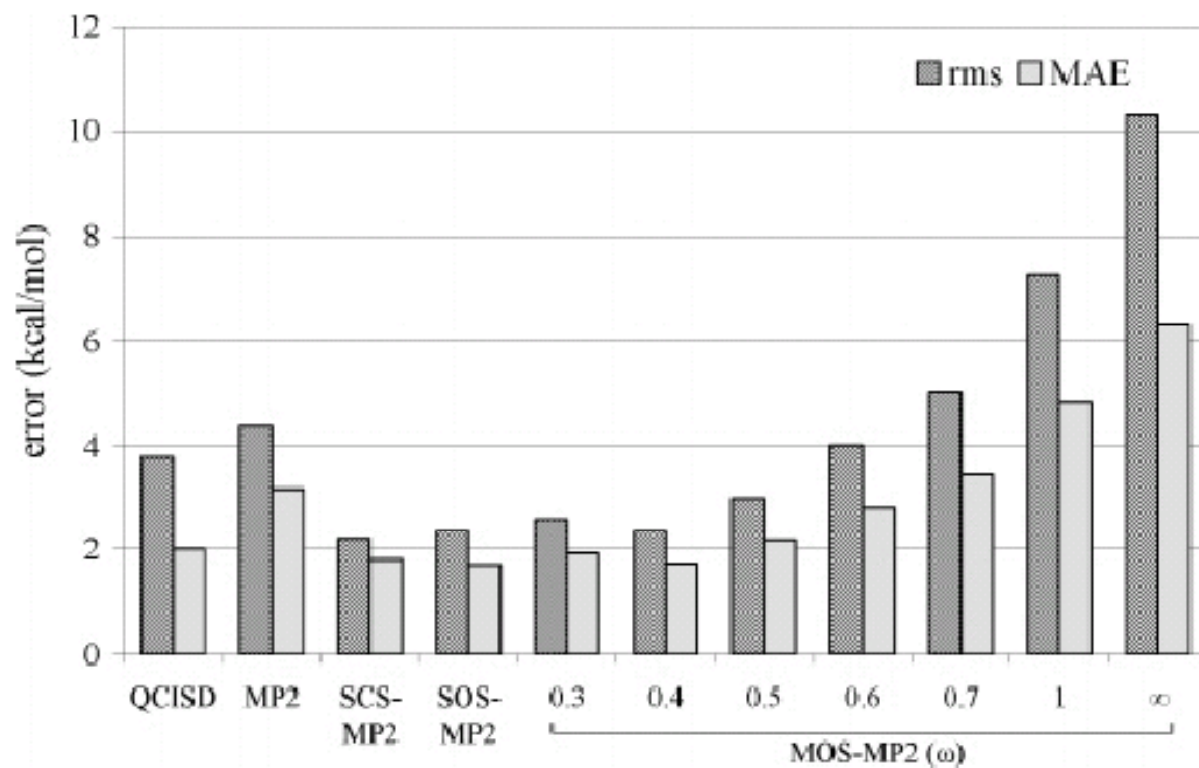
- Scaling factor that depends on the distance:

$$\frac{1}{r} + c_{\text{MOS}} \frac{\text{erf}(\omega r)}{r}$$

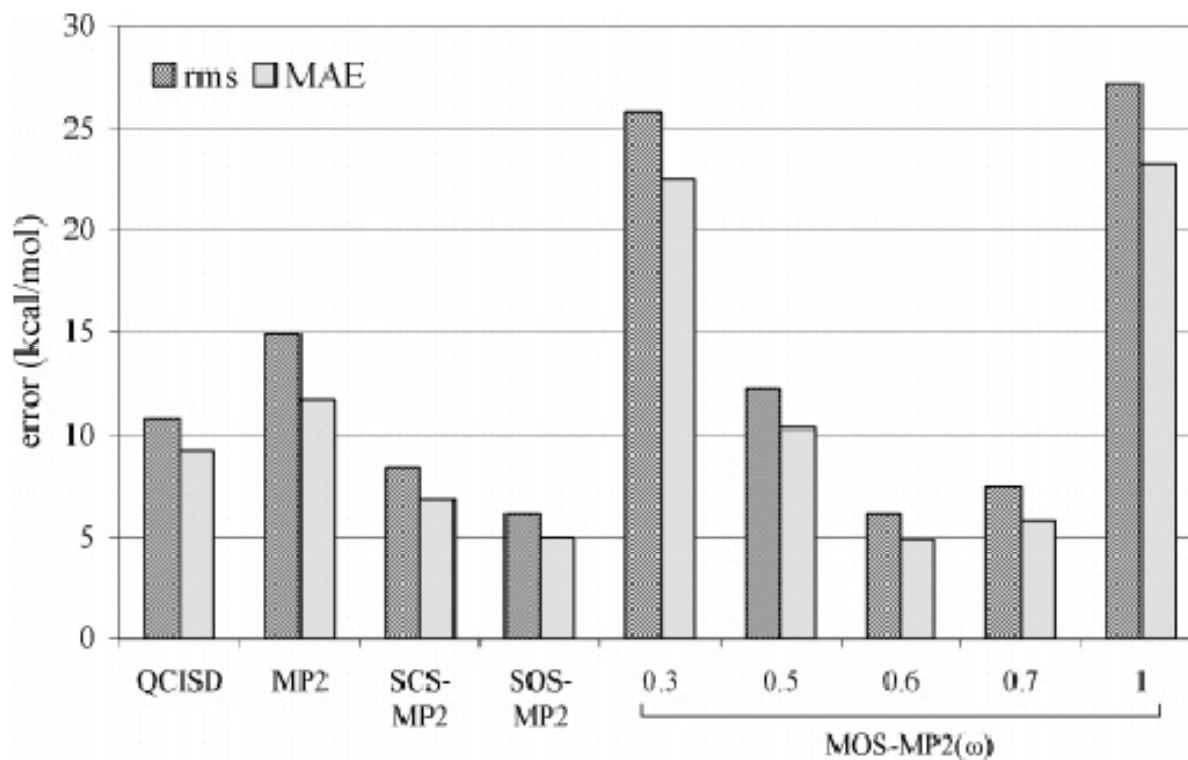
$$c_{\text{MOS}} = \sqrt{2} - 1$$

$$E_{\text{MP2}}^{\text{MOS}} \rightarrow 2E_{\text{MP2}}^{\text{OS}}, \quad r \rightarrow \infty$$

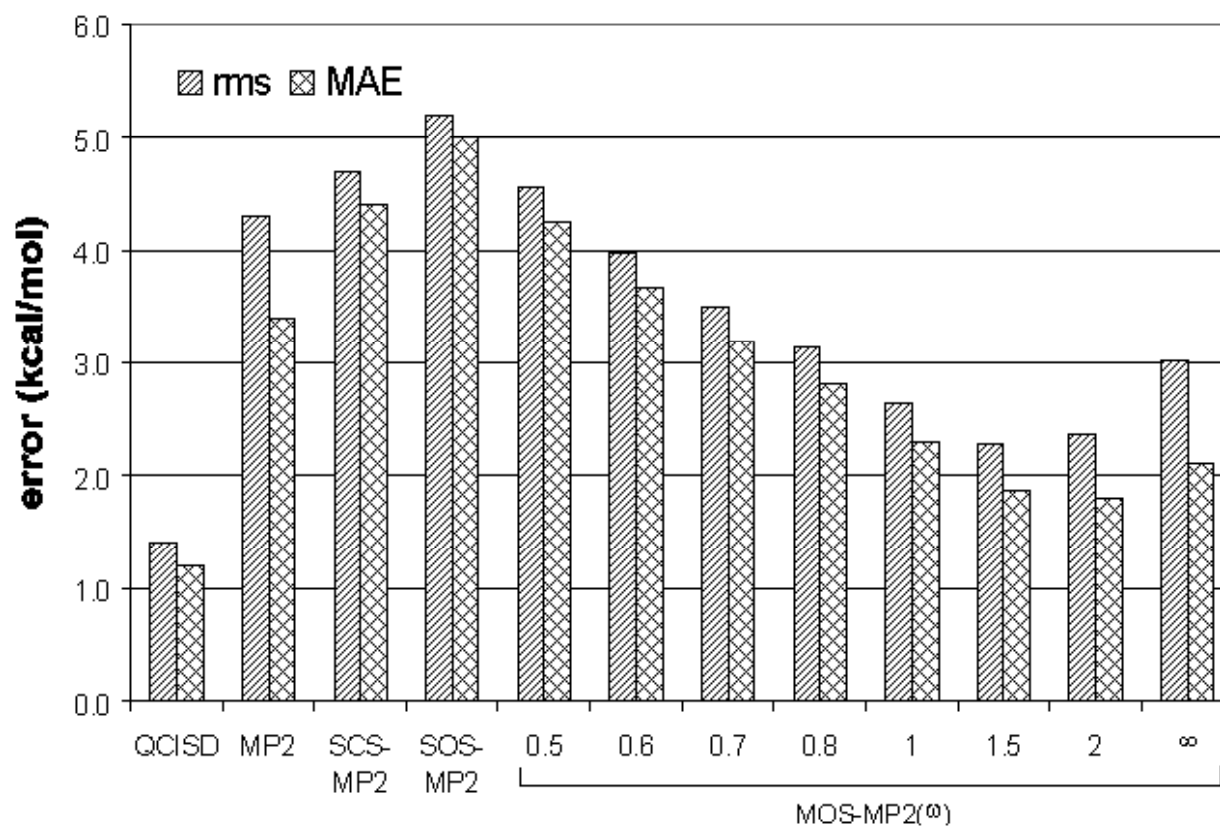
Reaction energies



Atomization Energies



Reaction Barriers



Dual-Basis MP2

- Avoid SCF convergence problem with diffuse basis functions

谢谢！